## Mole Calculation Worksheet

1) How many moles are in 15 grams of lithium?
2) How many grams are in 2.4 moles of sulfur?
3) How many moles are in 22 grams of argon?
4) How many grams are in 88.1 moles of magnesium?
5) How many moles are in 2.3 grams of phosphorus?
6) How many grams are in 11.9 moles of chromium?
7) How many moles are in 9.8 grams of calcium?
8) How many grams are in 238 moles of arsenic?

What are the molecular weights of the following compounds?
9) NaOH
12) $\mathrm{H}_{3} \mathrm{PO}_{4}$
10) $\mathrm{H}_{2} \mathrm{O}$
13) $\mathrm{Mn}_{2} \mathrm{Se}_{7}$
11) $\mathrm{MgCl}_{2}$
14) $\left(\mathrm{NH}_{4}\right)_{2} \mathrm{SO}_{4}$
15) How many grams are in 4.5 moles of sodium fluoride, NaF ?
16) How many moles are in 98.3 grams of aluminum hydroxide, $\mathrm{Al}(\mathrm{OH})_{3}$ ?
17) How many grams are in 0.02 moles of beryllium iodide, $\mathrm{Bel}_{2}$ ?
18) How many moles are in 68 grams of copper (II) hydroxide, $\mathrm{Cu}(\mathrm{OH})_{2}$ ?
19) How many grams are in 3.3 moles of potassium sulfide, $\mathrm{K}_{2} \mathrm{~S}$ ?
20) How many moles are in $1.2 \times 10^{3}$ grams of ammonia, $\mathrm{NH}_{3}$ ?
21) How many grams are in $2.3 \times 10^{-4}$ moles of calcium phosphate, $\mathrm{Ca}_{3}\left(\mathrm{PO}_{3}\right)_{2}$ ?
22) How many moles are in $3.4 \times 10^{-7}$ grams of silicon dioxide, $\mathrm{SiO}_{2}$ ?
23) How many grams are in 1.11 moles of manganese sulfate, $\mathrm{Mn}_{3}\left(\mathrm{SO}_{4}\right)_{7}$ ?

## Mole Calculation Worksheet - Answer Key

1) How many moles are in 15 grams of lithium? 0.46 moles
2) How many grams are in 2.4 moles of sulfur? 77.0 grams
3) How many moles are in 22 grams of argon? 0.55 moles
4) How many grams are in 88.1 moles of magnesium? 2141 grams
5) How many moles are in 2.3 grams of phosphorus? 0.074 moles
6) How many grams are in 11.9 moles of chromium? 618.8 grams
7) How many moles are in 9.8 grams of calcium? $\mathbf{0 . 2 4}$ moles
8) How many grams are in 238 moles of arsenic? 17,826 grams

What are the molecular weights of the following compounds?
9) $\mathrm{NaOH} \quad 40.1$ grams
10) $\mathrm{H}_{2} \mathrm{O} \quad 18.0$ grams
11) $\mathrm{MgCl}_{2} \quad 95.3$ grams
12) $\mathrm{H}_{3} \mathrm{PO}_{4} \quad 98.0$ grams
13) $\mathrm{Mn}_{2} \mathrm{Se}_{7} \quad 663.0$ grams
14) $\left(\mathrm{NH}_{4}\right)_{2} \mathrm{SO}_{4} \quad 132.1$ grams
15) How many grams are in 4.5 moles of sodium fluoride, NaF? 189 grams
16) How many moles are in 98.3 grams of aluminum hydroxide, $\mathrm{Al}(\mathrm{OH})_{3}$ ? 1.26 moles
17) How many grams are in 0.02 moles of beryllium iodide, $\mathrm{Bel}_{2}$ ? 5.2 grams
18) How many moles are in 68 grams of copper (II) hydroxide, $\mathrm{Cu}(\mathrm{OH})_{2}$ ? 0.70 moles
19) How many grams are in 3.3 moles of potassium sulfide, $\mathrm{K}_{2} \mathrm{~S}$ ? 364.0 grams
20) How many moles are in $1.2 \times 10^{3}$ grams of ammonia, $\mathrm{NH}_{3}$ ? 70.6 moles
21) How many grams are in $2.3 \times 10^{-4}$ moles of calcium phosphate, $\mathrm{Ca}_{3}\left(\mathrm{PO}_{3}\right)_{2}$ ? 0.064 grams
22) How many moles are in $3.4 \times 10^{-7}$ grams of silicon dioxide, $\mathrm{SiO}_{2}$ ? $5.66 \times 10^{-9}$ moles
23) How many grams are in 1.11 moles of manganese sulfate, $\mathrm{Mn}_{3}\left(\mathrm{SO}_{4}\right)_{7}$ ? 929.5 grams

